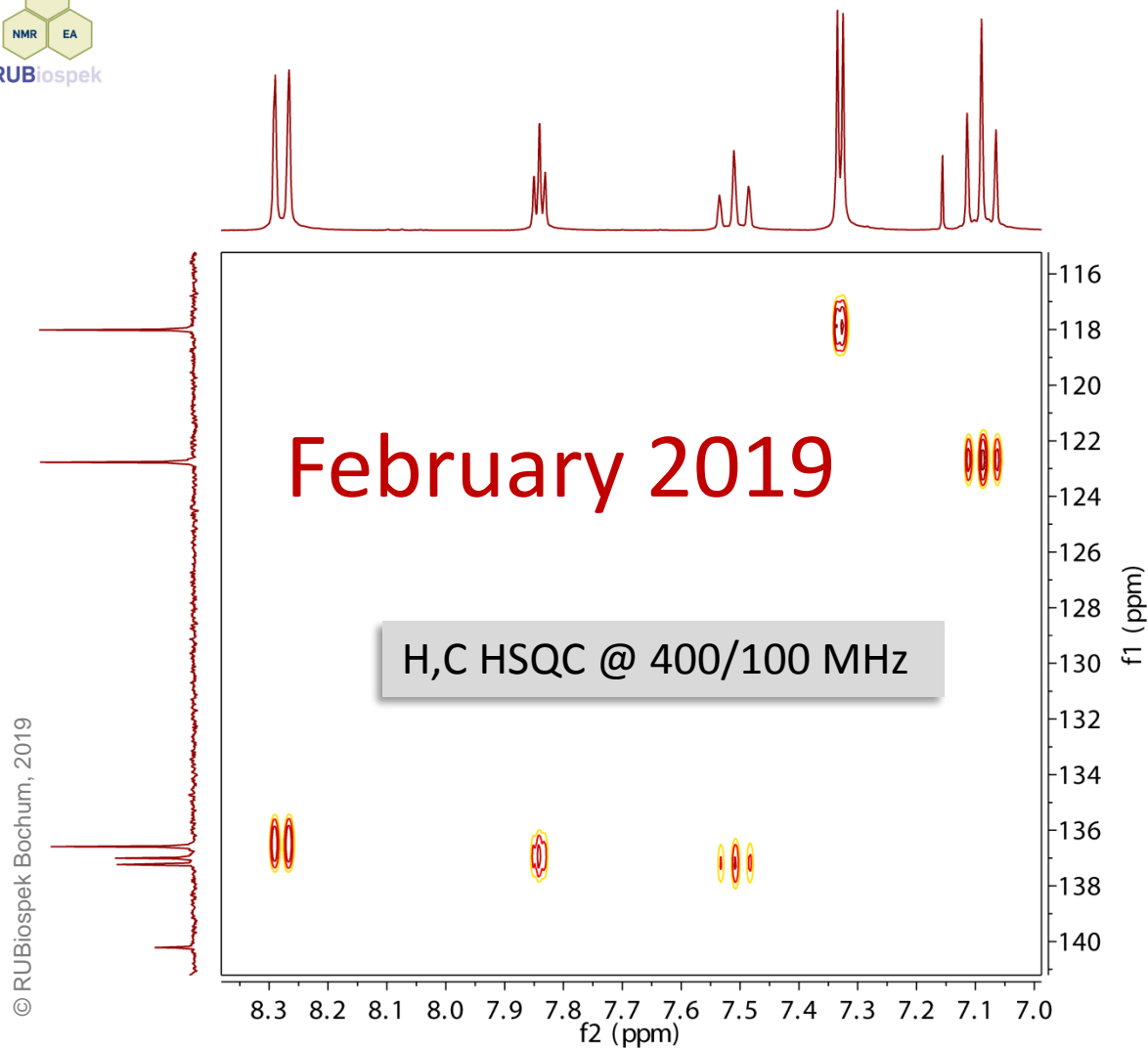


Problem of the Month:



Problem of the Month:

February 2019

Strategy

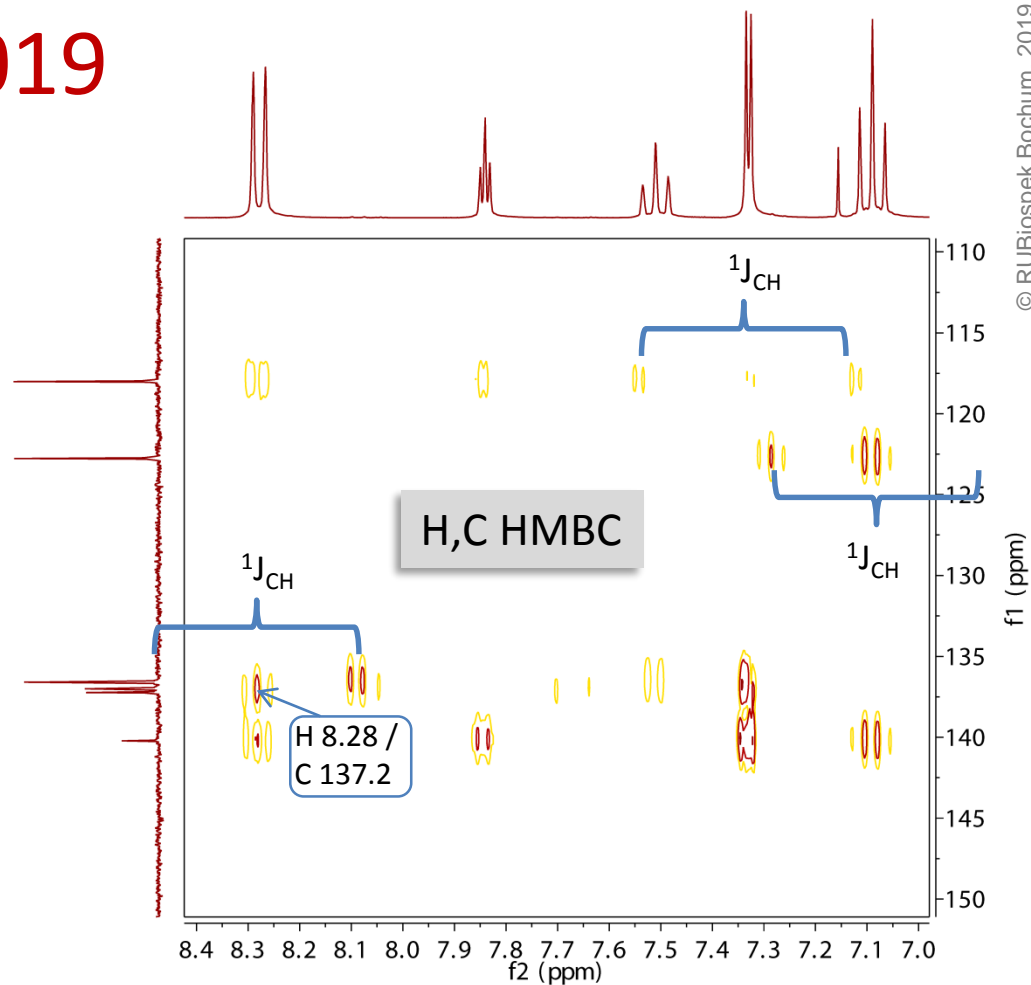
- (1) The degree of unsaturation (DBE) can be derived from the molecular formula: $C_{10}H_8$.
- (2) Check the number of distinct signals for carbon and hydrogen NMR: This will give you a clue.
- (3) Carefully analyse the structure of the multiplets visible in the 1D 1H NMR spectrum - if you are struggling, there is a H,H COSY provided, too.

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^{13}C chemical shifts:
140.2, 137.2, 137.0,
136.6, 122.8, 118.0

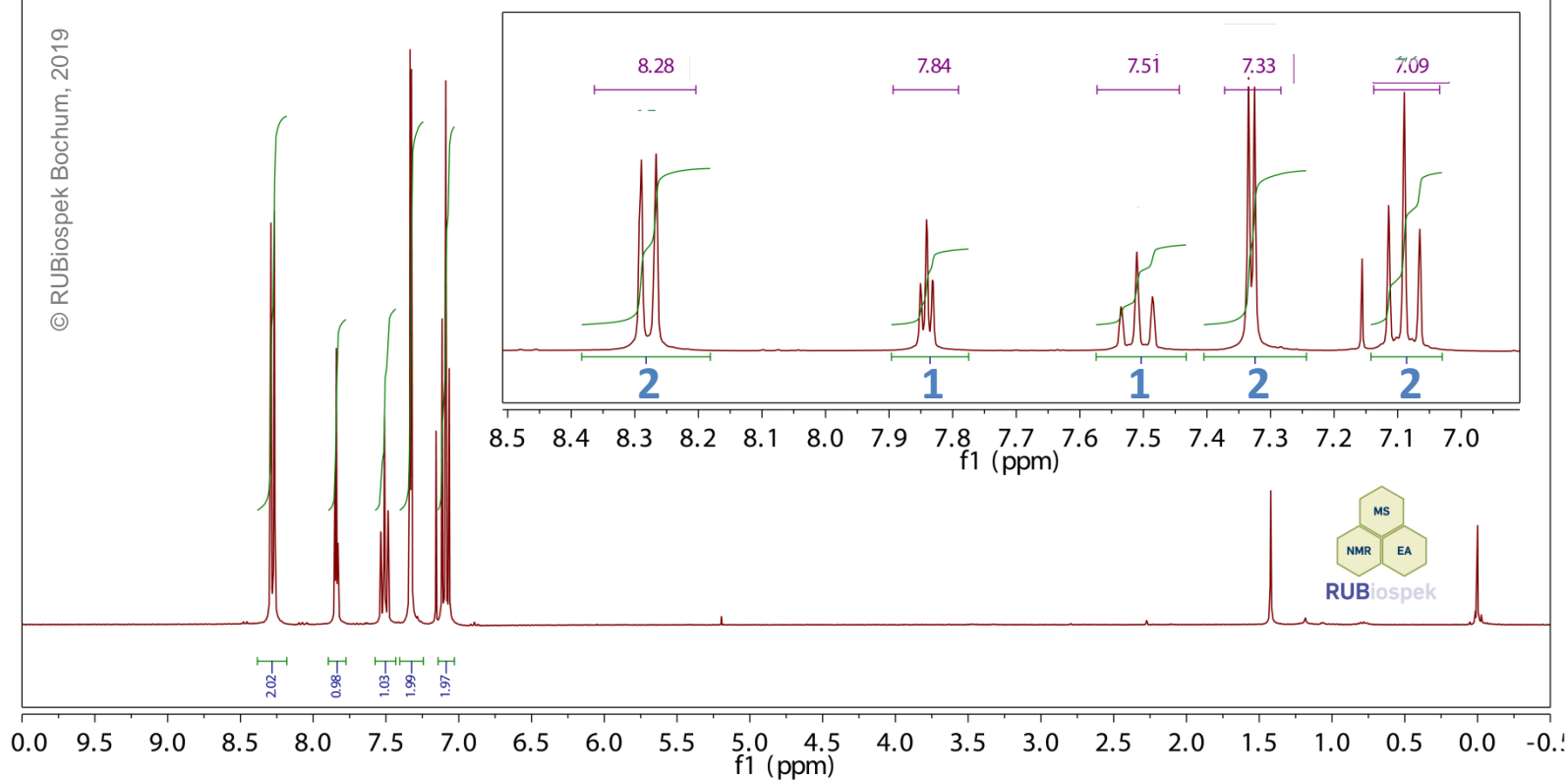
^1H chemical shifts:
8.28, 7.84, 7.51,
7.33, 7.09



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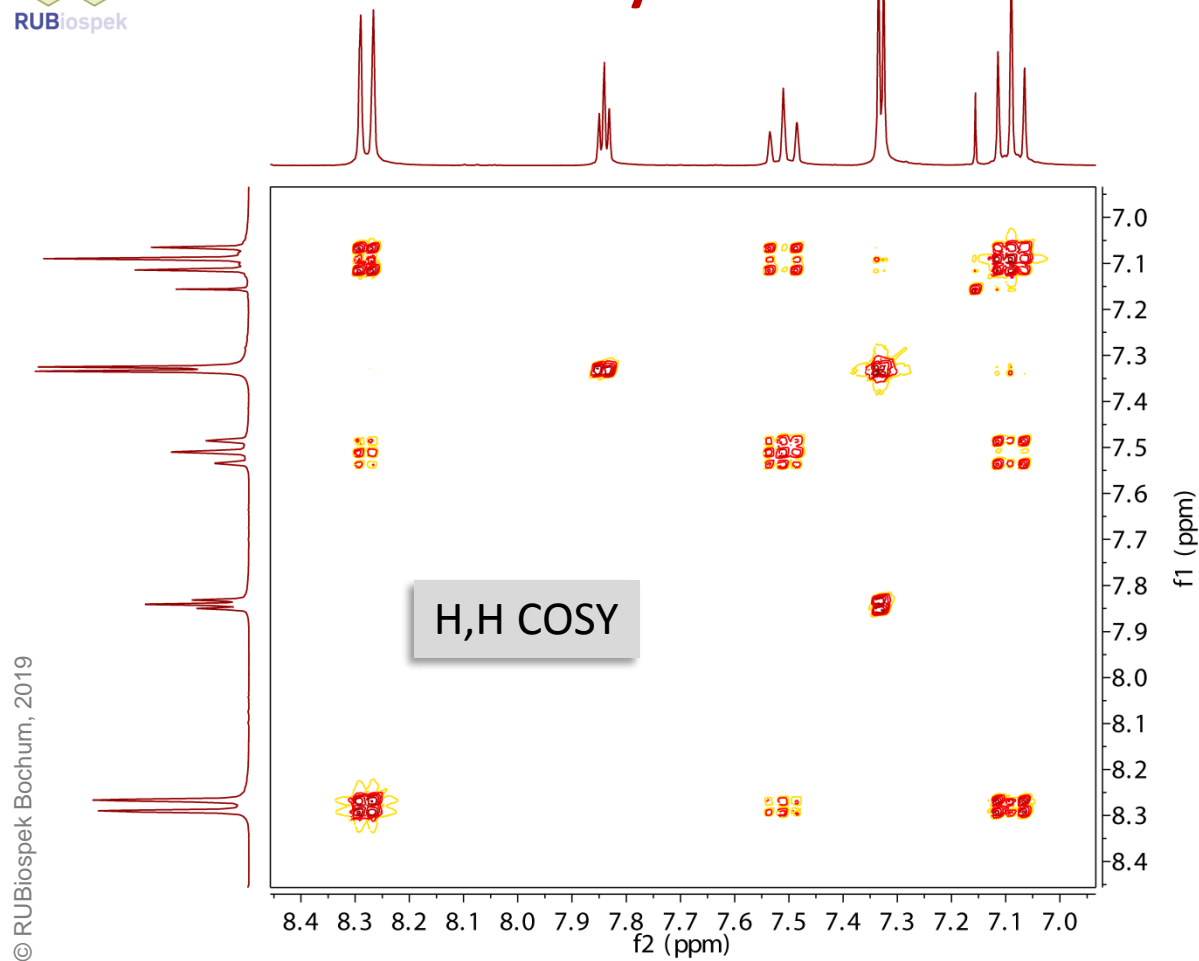
^1H NMR spectrum
recorded at 400 MHz



Problem of the Month:



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Problem of the Month:

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Hints

- (1) It is most important to recognise the symmetry of the compound - all chemically equivalent atoms give rise to one common signal. Integration of proton signals is given below the corresponding peaks.
- (2) In the HMBC, most of the cross peaks derive from ${}^3J_{\text{CH}}$ coupling. In the COSY experiment, two spin systems may be identified - caution, there are also long range correlations visible ($> {}^3J_{\text{HH}}$).

Problem of the Month:

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Solution

- (1) The molecule of February is quite symmetric - there are four positions that exists twice. Also, it is an aromatic compound for the chemical shifts in the proton spectrum, it contains 7 DBE.
- (2) The two spin systems are divided into an AX_2 and an $AA'MM'X$ part. You may simulate such a 5-spin system at <http://nmr.cheminfo.org> (middle box in the ,tools' section, choose ABCDE system and set δ 8.28 for A/B, 7.51 for C and 7.09 for D/E. Providing uniform 3J couplings of 7 Hz between A/D, B/E, C/D and C/E is sufficiently accurate).
- (3) Connecting the annulated system is feasible through HMBC correlations. If you are still in doubt - enter the chemical shifts of the ^{13}C signals (compare results with and without carbon multiplicities, i.e. S = no proton attached, D = proton attached) in nmrshiftdb2's search function.