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And now the challenge

With three chemical shifts and two coupling constants, <u>you can simulate</u> the three multiplets in the proton spectrum.

The simulation for the multiplet at about 3.4 ppm will provide an almost perfect result, but in the case of the other two multiplets there are small but significant deviations: The multiplet at about 1.9 ppm looks a bit more complex than expected, although there exists a clear base structure. But the multiplet at about 2.4 ppm should be a pure triplet.

Because of the small difference in chemical shift between the multiplets at 1.9 ppm and 2.4 ppm, fine splitting is visible in the simulation for which the experimental resolution is insufficient.

But the real issue are the two "warts" in the experimental spectrum, for which no evidence is visible in the simulation.

Hint:

To answer this extremely difficult question, trawl through the literature for the proton spectrum of 1-bromo-2chloroethane, marvel a little bit about what you find, and try to understand the theoretical explanation of this spectrum.



Contributions

